CLASS: 9 NAME OF THE STUDENT:

I)Answer the following questions

1. If Z=5, What would be the valency of the of the element?

2. What is the atomic number and number of neutrons in $_{19}$ K³⁹

3. Frame two questions to understand "Homogeneous mixture"

4.3g of Hydrogen reacts with 24g of Oxygen to produce 27g of Water. Which law supports the given information

II) Answer the following questions.

5.Complete the following table

Anions 🔶	Chloride	Hydroxide
Potassium		
Zinc		

6. What are the applications of isotopes

III) Answer the following questions.

7. Write the postulates of Bohr's model of an atom

Write the steps you would use for making tea. Use the following words given below and write the steps for making tea.

(OR)

solution, solvent, solute, dissolve, soluble, insoluble, filtrate and residue

8. How do you separate two miscible liquids whose boiling points difference is more than 25°C. Draw a neat diagram of the process?

(OR)

Write the law of conservation of mass. Draw the diagram to show the experimental setup for the law of conservation of mass

IV)Choose the correct answer from the given choices				8x1/2=4M	
9. Radha : Salt is a c	compound				
Rajani : Salt is a n	nixture. Who is	correct?		()
A)Radha	B)Rajani	C) Both	D) None of these		
10.Which of the following is a colloidal dispersion?			()	
A)salt solution	B)mixture of	oil and water	C)shoe- polish D)sand and water		
11.Tyndall Effect ca	annot be shown b	У		()
A)colloids	B)suspensions	C)emulsions	s D)solutions		
12.Avogadro const				()
A)6.022x10 ²⁰	B)6.022x10 ²¹	C)6.022×	(10 ²² D)6.022x10 ²³		
13.Molecular mass	of H ₂ SO ₄ is			()
A)18u	B)48u	c)98u	D)100u		
14.The common name of hydrogen monoxide is			()	
A) water	B)salt C)wa	shing soda	D)Baking soda		
15.The property of having 8 electrons in the outer most orbit is called			()	
A)Atomicity	B)Valency	C)Octet	D)Inertness		
16.The charge of alpha particle is			()	
A) positive	B)negative	C)neutral	D)no charge		

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Max.Marks:20

ROLL NO:_

4X1=4M

2x2=4M

2x4=8M